

7th and 8th Grade Science Checklist for Days 1-10

Day 1

7th Grade: "Motion" Crossword _____

8th Grade: Page 1 of "Cooking with Elements" _____

Day 2

7th Grade: "Speed, Velocity, and Acceleration" Crossword _____

8th Grade: Page 2 of "Cooking with Elements" _____

Day 3

7th Grade: Page 1 "Speed Machine" _____

8th Grade: Page 1 of "Counting Elements" _____

Day 4

7th Grade: Page 2 "Speed Machine" _____

8th Grade: Page 2 of "Counting Elements" _____

Day 5

7th and 8th Grade: Human impact on the Environment and Conservation Seek & Find Activity _____

Day 6

7th and 8th Grade: Newton's First and Second Laws of Motion _____

Day 7

7th and 8th Grade: Newton's Third Law of Motion _____

Day 8

7th and 8th Grade: Page 1 of Balanced VS Unbalanced Forces _____

Day 9

7th and 8th Grade: Page 2 of Balanced VS Unbalanced Forces _____

Day 10

7th and 8th Grade: Lab Safety Review _____

Use for 8th Grade Days 1+2

Periodic Table of the Elements

Number	Symbol	Name	Mass
1	H	Hydrogen	1.008
2	He	Helium	4.003
3	Li	Lithium	6.941
4	Be	Beryllium	9.012
5	B	Boron	10.811
6	C	Carbon	12.011
7	N	Nitrogen	14.007
8	O	Oxygen	15.999
9	F	Fluorine	18.998
10	Ne	Neon	20.180
11	Na	Sodium	22.990
12	Mg	Magnesium	24.305
13	Al	Aluminum	26.982
14	Si	Silicon	28.086
15	P	Phosphorus	30.974
16	S	Sulfur	32.066
17	Cl	Chlorine	35.453
18	Ar	Argon	39.948
19	K	Potassium	39.098
20	Ca	Calcium	40.078
21	Sc	Scandium	44.956
22	Ti	Titanium	47.887
23	V	Vanadium	50.942
24	Cr	Chromium	51.996
25	Mn	Manganese	54.938
26	Fe	Iron	55.845
27	Co	Cobalt	58.933
28	Ni	Nickel	58.693
29	Cu	Copper	63.546
30	Zn	Zinc	65.38
31	Ga	Gallium	69.723
32	Ge	Germanium	72.631
33	As	Arsenic	74.922
34	Se	Selenium	78.971
35	Br	Bromine	79.904
36	Kr	Krypton	83.798
37	Rb	Rubidium	85.468
38	Sr	Strontium	87.62
39	Y	Yttrium	88.906
40	Zr	Zirconium	91.224
41	Nb	Niobium	92.906
42	Mo	Molybdenum	95.95
43	Tc	Technetium	98.907
44	Ru	Ruthenium	101.07
45	Rh	Rhodium	102.906
46	Pd	Palladium	106.42
47	Ag	Silver	107.868
48	Cd	Cadmium	112.414
49	In	Indium	114.818
50	Sn	Tin	118.710
51	Sb	Antimony	121.760
52	Te	Tellurium	127.6
53	I	Iodine	126.904
54	Xe	Xenon	131.293
55	Cs	Cesium	132.905
56	Ba	Barium	137.328
57-71	Lanthanide Series		
72	Hf	Hafnium	178.49
73	Ta	Tantalum	180.948
74	W	Tungsten	183.84
75	Re	Rhenium	186.207
76	Os	Osmium	190.23
77	Ir	Iridium	192.222
78	Pt	Platinum	195.085
79	Au	Gold	196.967
80	Hg	Mercury	200.592
81	Tl	Thallium	204.383
82	Pb	Lead	207.2
83	Bi	Bismuth	208.980
84	Po	Polonium	[209]
85	At	Astatine	208.980
86	Rn	Radon	222.018
87	Fr	Francium	223.020
88	Ra	Radium	226.025
89-103	Actinide Series		
104	Rf	Rutherfordium	[261]
105	Db	Dubnium	[262]
106	Sg	Seaborgium	[266]
107	Bh	Berkelium	[264]
108	Hs	Hassium	[265]
109	Mt	Moscovium	[270]
110	Ds	Darmstadtium	[281]
111	Rg	Roentgenium	[280]
112	Cn	Copernicium	[285]
113	Nh	Nihonium	[286]
114	Fl	Flerovium	[289]
115	Mc	Moscovium	[289]
116	Lv	Livermorium	[293]
117	Ts	Tennessine	[294]
118	Og	Oganesson	[294]
119	Uue	Ununennium	[295]
120	Uub	Unbium	[296]
121	Uut	Untrium	[297]
122	Uuq	Unquadium	[298]
123	Uuh	Unhexium	[299]
124	Uuq	Unseptium	[300]
125	Uuh	Unoctium	[301]
126	Uub	Unnium	[302]
127	Uut	Untrium	[303]
128	Uuq	Unquadium	[304]
129	Uuh	Unhexium	[305]
130	Uub	Unnium	[306]
131	Uut	Untrium	[307]
132	Uuq	Unquadium	[308]
133	Uuh	Unhexium	[309]
134	Uub	Unnium	[310]
135	Uut	Untrium	[311]
136	Uuq	Unquadium	[312]
137	Uuh	Unhexium	[313



Name _____

Date _____

8th Grade

* Day 1 - Complete

1-5

Activity 1 4: Cooking with the Elements

* Day 2 - Complete

6-10

Directions: For each element combination in parentheses below, use the symbols for the elements to obtain a scrambled word. Then unscramble the letters to form the correct words. Write the symbols in the answer blank following each group of elements. This will help you complete each numbered paragraph.

Example: (boron, indium, oxygen, tantalum) = BInOTa, which unscrambles to form the word OBTAIn.

- For breakfast we (yttrium + francium) _____ eggs, (cobalt + nitrogen + barium) _____ and (hydrogen + hydrogen + arsenic) _____ (oxygen + nitrogen + tungsten + bromine) _____ potatoes, and toast (astatine + tungsten + helium) _____ or (hydrogen + tellurium + tungsten + iodine) _____ bread. Or, we can have (nitrogen + calcium + einsteinium + protactinium + potassium) _____ or waffles and sausage, or (aluminum + cerium + rhenium) _____ such as (radon + cobalt) _____ (lanthanum + potassium + fluorine + einsteinium) _____ (nitrogen + iodine + silicon + radium) _____ (boron + nitrogen + radium) with milk. _____
- (thorium + helium + aluminum + _____ yttrium)(potassium + actinium + sulfur tin) _____ would be fruits, such as (sodium + sodium + barium + sulfur) _____, grapes, (sulfur + iodine + tungsten + potassium + iodine) apples, and oranges and different (einsteinium + carbon + helium + einsteinium) _____ and (potassium + chromium + erbium + actinium + sulfur) _____ Of course, most of us would (erbium + radium + thorium) have (hydrogen + phosphorus + sulfur + carbon + iodine) (iodine + oxygen + cobalt + potassium + einsteinium) _____ or (nitrogen + dysprosium + calcium) _____
- For drinks, we (fluorine + phosphorus + rhenium + erbium) (calcium + cobalt + lanthanum + cobalt) _____ or another type of soda (vanadium + erbium + oxygen) _____ milk, juice or (erbium + astatine + tungsten) _____
- Most people have fast food and (selenium + _____ uranium) the drive (ruthenium + sulfur + thorium) _____ for lunch. They usually have only half an hour and (oxygen + carbon + selenium + holmium) _____ (carbon + tantalum + osmium) _____ or hamburgers and French (einsteinium + iodine + francium) _____

Sometimes they will be (carbon + yttrium + lutetium + _____
 potassium) and have a salad, (uranium + phosphorus + _____, sulfur +
 oxygen) sandwich, or (neon + iodine + _____ hydrogen + selenium +
 carbon) take-out. At (erbium + oxygen + thorium) _____ times, people, especially
 students, eat (holmium + _____ sulfur + carbon + sodium) _____ or (carbon + lithium
 + iodine + hydrogen) _____ cheese (iodine + francium + einsteinium)
 _____.

8th Gr.

Days 1+2

(continued)

Name _____

Date _____



Activity 14: Cooking with the Elements (continued)

5. Dinners are the big meals. (iodine + sulfur + _____ thorium) is (helium
 + tungsten + nitrogen) _____ families (thorium + _____ gallium +
 erbium) _____ together after a long day. Dinners usually consist of a main dish containing
 some type of meat. The meat can be (neon + _____, terbium
 + oxygen) (americium + hydrogen) _____, pork (sulfur + _____
 phosphorus + carbon + holmium) chicken, (boron + barium + yttrium) _____
 (carbon + barium + potassium) ribs, prime rib, or (iodine + hydrogen + sulfur + fluorine)
 _____.

6. Of course, there is always some type of carbohydrate. (iodine + thorium + sulfur)
 _____ is usually a potato, which we can bake, mash, (yttrium + francium)
 _____, scallop, or boil. For variety, there is also rice or (tantalum + arsenic +
 phosphorus) _____.

7. There usually is a (holmium + cerium + iodine + carbon) of vegetables.

Some (sulfur + carbon + iodine + holmium + _____ cerium) are (radon +
 cobalt) _____, peas, (cobalt + lithium + oxygen + bromine + carbon) beans,
 _____.

(silver + arsenic + phosphorus + uranium + argon + sulfur) or squash.

8. One of my favorite (sulfur + uranium + sulfur + oxygen + phosphorus) is

(tungsten + neon) _____ England (americium + chlorine) _____ chowder. I

8th
Grade
Days 1+2

(gold + sulfur + tellurium) _____ the (nitrogen + barium + cobalt) _____
and (nitrogen + oxygen + nickel + oxygen + sulfur) _____ first. Then I add (astatine +
tungsten + erbium) _____, (sulfur + chlorine + americium) _____, celery,
and (iodine + sulfur + sulfur + phosphorus + cerium) _____ such as (yttrium + barium)
_____ leaf, thyme, and marjoram. The diced potatoes and (rhenium + carbon +
americium) _____ are added about thirty minutes (oxygen + rhenium + beryllium +
fluorine) _____ serving.

9. The best part is dessert. There are many different (potassium + calcium + einsteinium)
_____ and (iodine + einsteinium + _____ phosphorus)
(uranium + rubidium + rhodium + barium) _____ looks like red celery, and is
tart (helium + tungsten + nitrogen) _____ baked in (phosphorus + einsteinium +
iodine) _____

Another simple dessert is (cerium + _____ iodine)(americium + carbon +
rhenium) _____ (iodine + _____ thorium + sulfur) can be
served with (erbium + oxygen + thorium) _____ desserts or
(yttrium + boron) itself, in a (neon + cobalt) _____
_____ or a dish, (iodine + thorium + tungsten) (cobalt + tellurium + carbon +
lanthanum + holmium) or (yttrium + neodymium + calcium) _____ toppings.

10. A fancy dessert is (phosphorus + rhenium + carbon + _____
einsteinium) with a _____
(neodymium + boron + radium + yttrium) sauce. Many (neon + _____ iodine +
fluorine) _____ restaurants will (cerium + lanthanum) their desserts with
(nitrogen + boron + dysprosium + radium) to make a flaming dessert.

HOW TO COUNT ATOMS

*Notes for 8th
Grade to help
with "Counting
Elements"
Activity
on Days
3 & 4

Background Information:

- **subscripts** – the little numbers that tell how many atoms there are (ex: In $3\text{H}_2\text{O}$, the 2 is the subscript)
- **coefficients** – regular-sized numbers that tell how many molecules there are (ex: In $3\text{H}_2\text{O}$, the 3 is the coefficient)

Example: $3\text{H}_2\text{O}$

- The subscript 2 in the example above comes after the H.
This means there are **two H's** (hydrogen atoms) in **each** molecule.
- The coefficient 3 shows that there are **three** of the H_2O molecules.

We could draw each molecule to help show us how many atoms are present.



Now we can count the number of H's and the number of O's.

$3\text{H}_2\text{O}$ has **6 hydrogen atoms** and **3 oxygen atoms**.

However, drawing out each equation is **NOT PRACTICAL**. The number of atoms can be counted without having to draw it out.

To find out the number of atoms: **MULTIPLY** all the **SUBSCRIPTS** in the molecule by the **COEFFICIENT**. (This will give you the number of atoms of each element.)

To mathematically find the number of elements that make up $3\text{H}_2\text{O}$, we multiply the 2 by the coefficient 3 to find that there are 6 H's. Then we multiply the 1 by the coefficient 3 to find that there are 3 O's.

NOTE: Although the 1 is usually not written, $3\text{H}_2\text{O}$ can be written as $3\text{H}_2\text{O}_1$.

(In other words, $3\text{H}_2\text{O}$ and $3\text{H}_2\text{O}_1$ are the **same thing**.)

HOW TO COUNT ATOMS IN A CHEMICAL FORMULA

(5 Easy Steps)

Step 1: Write the chemical formula

Step 2: List all the atoms

Step 3: Count the number of atoms of each element in 1 molecule.

Step 4: Multiply the number of atoms of each by the coefficient.

Step 5: Make sure your answer makes sense.

Practice Examples



C - $2 * 6 = 12$

H - $6 * 6 = 36$

O - $1 * 6 = 6$



C - $6 * 4 = 24$

H - $12 * 4 = 48$

O - $6 * 4 = 24$



Ca - $1 * 5 = 5$

O - $2 * 5 = 10$

H - $2 * 5 = 10$

Counting Atoms and Elements

Counting Atoms Notes

C

CO₂

Mg(OH)₂

3(OH)₂

8th Grade
Day 3

Counting Atoms and Elements

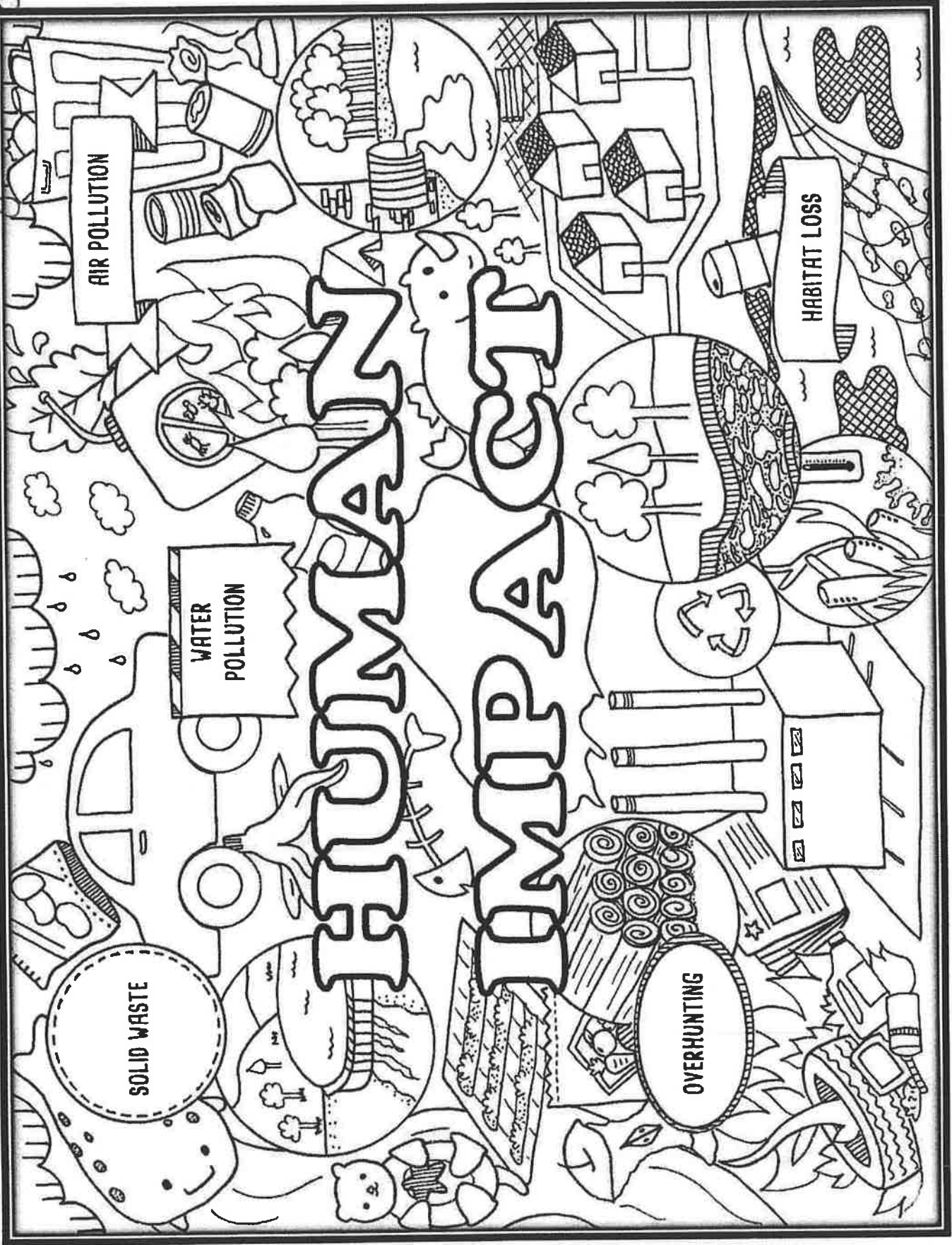
8th Grade

Day 4

Counting Atoms and Elements Practice

	# of Atoms	# of Elements
P	_____	_____
SO ₄	_____	_____
3O ₂	_____	_____
H ₃ PO ₄	_____	_____
2Zn(OH) ₂	_____	_____
Na ₂ SiO ₃	_____	_____
Mg(NO ₃) ₂	_____	_____
Ba ₃ (PO ₄) ₂	_____	_____

7/5/17 8:30 AM Grades - Days 5



7th/8th
Day 5

NAME _____

SEEK & FIND SCIENCE- HUMAN IMPACT ON THE ENVIRONMENT

SEEK & FIND

1. CAUSES OF AIR POLLUTION (3)
2. CAUSES OF WATER POLLUTION (5)
3. CAUSES OF HABITAT LOSS (5)
4. ANIMALS THAT ARE OVERHUNTED (4)
5. LANDFILL
6. RECYCLING SYMBOL
7. ITEMS THAT CAN BE RECYCLED (5)
8. ITEMS THAT CAN BE COMPOSTED (3)
9. CLEAR CUTTING
10. FOREST FIRE
11. ACID RAIN
12. HARVESTING NATURAL RESOURCE (2)
13. BURN PILE
14. OIL SPILL
15. CORAL BLEACHING
16. USES FOSSIL FUELS (3)
17. CAUSE OF ACID RAIN (3)

DEFINE KEY TERMS

- ☐ SOLID WASTE- _____
- ☐ WATER POLLUTION- _____
- ☐ AIR POLLUTION- _____
- ☐ OVER HUNTING- _____
- ☐ HABITAT LOSS- _____

COLOR & SHARE!

#SEEKANDFINDSCIENCE

Day 6 7th / 8th Grades

Name _____

NEWTON'S FIRST & SECOND LAWS

Directions: Read the information below.

Newton's First Law of Motion

An object at rest will stay at rest and an object in motion will stay in motion until acted upon by an outside force. The first law of motion is all about inertia. This may seem complex, but it's actually pretty easy to understand. If you're sitting in your seat, you don't expect to start moving across the classroom. Nothing is making you move, right? You set your homework paper down on your bed and decide to go outside. When you come back in those papers should still be there. These objects (you in the chair and your homework on the bed) are at rest, meaning they are not moving. They're still. They will continue to be at rest unless something makes them move – in other words, they're acted upon by an outside force. If the wind were to blow through a window in your room, that would be a force that could move your papers. Likewise, if somebody bumped into your seat you would expect it to begin moving. No object at rest will ever begin moving on its own without the help of an outside force.

Objects that are in motion will also stay in motion unless a force compels them to stop. A roller coaster may be gliding along a track, but when it reaches a hill it will slow down because the force of gravity wants to pull it back down. A football player who throws a football hopes that it will fly through the air as far as possible. However, a variety of factors including gravity and wind will act against that motion and cause the football to eventually fall. A quarterback who throws for a touchdown hopes that the football does not get acted up by the outside force of an interception, preventing it from continuing to travel down the field. A car can coast along a road, but will eventually slow down if the accelerator is not applied because of the force of friction.

Physicists like to say that objects which have a net force of 0 will have no change in motion. As you stand on the ground you exert a force on the ground, but the ground also pushes back up onto you. The two forces are equal which means there is no net force, thus the object (you) stays still. If you pushed on the ground with a greater force, then the ground would cave in downward. If the ground pushed back up on you with a greater force, then you would rise upward.

Newton's Second Law of Motion

This law states that force is equal to mass times acceleration.

$$\text{Force} = \text{mass} \times \text{acceleration}$$

In other words, we know that if an object's acceleration or mass changes so will its force. It may not seem clear, but this is actually very logical. Imagine a student walking down the hall to class. Hands are empty, backpack is on. If you run into that student it would probably hurt, but not much. Now, assume that student is carrying a heavy load of books. It now requires more effort (force) for that student to maintain the same speed (acceleration) walking down the hall. That's because the mass has increased from the addition of the books. If that student was walking down the hall with empty hands again and increased his or her speed by walking faster, it would hurt more to run into them. This time it's because the acceleration increased.

If Newton's first law of motion explains how objects do not change their total net force, the second law explains how they do change their total force.

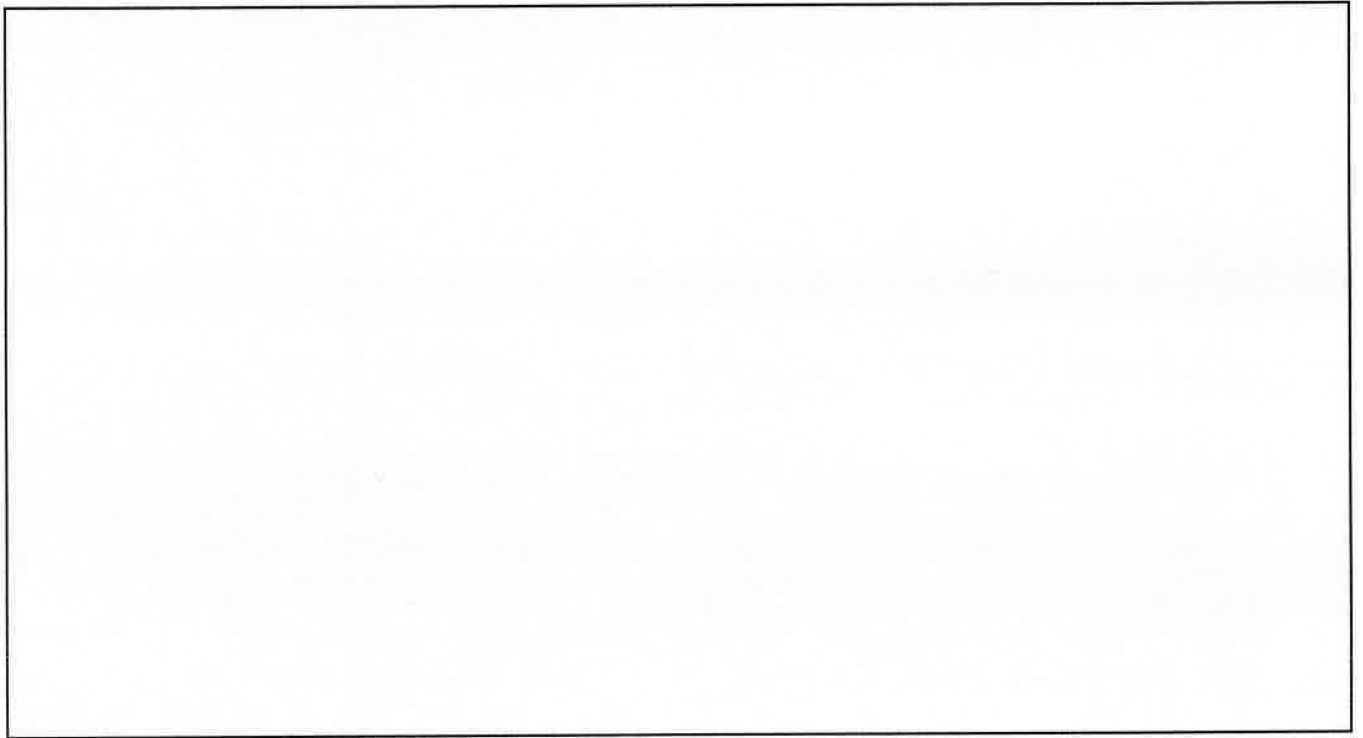
Directions: Answer the questions below.

1. If a hockey puck is traveling across the ice, it will continue moving at the same speed forever unless acted upon by an outside force. If no humans or hockey sticks touch the puck, what force might slow it down?
2. If a large and small player collide, will one or both players have a force exerted upon them? Explain your answer using Newton's Second Law of Motion.
3. The amount of force an object has is a product of what two factors?

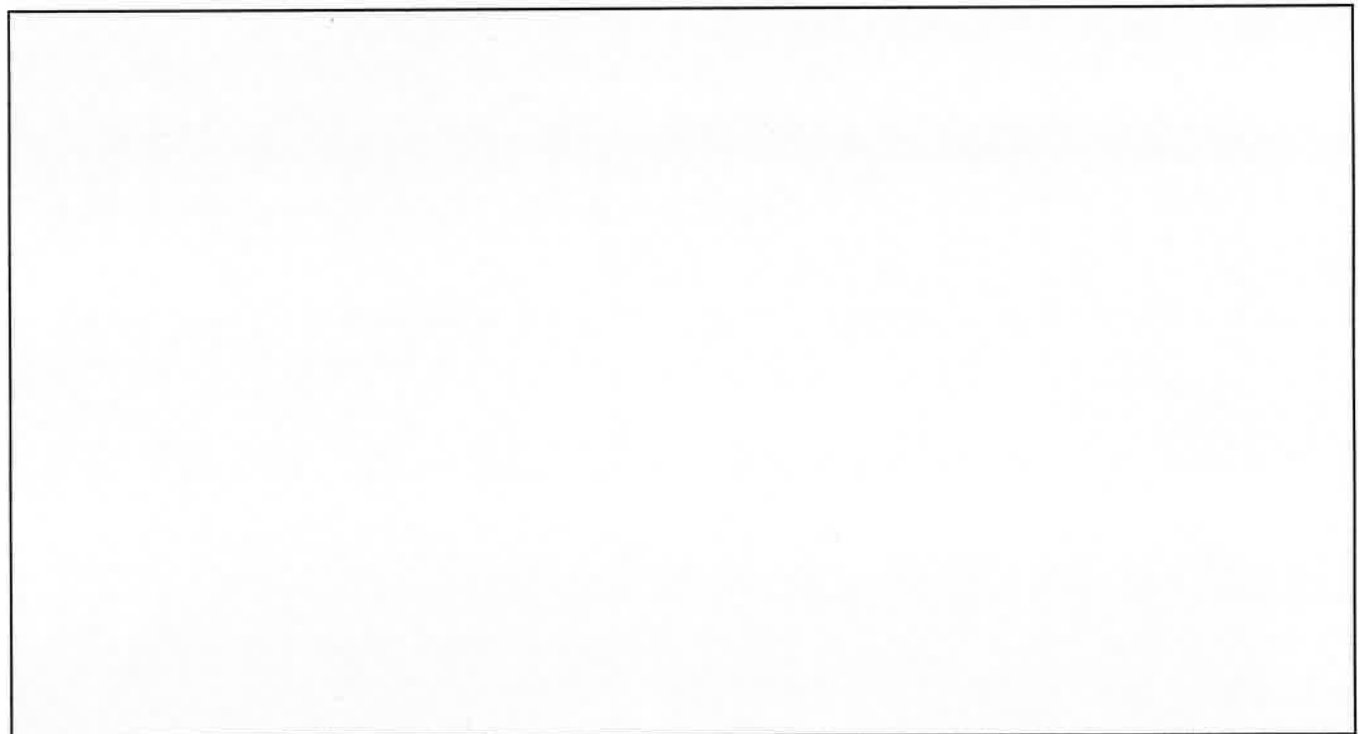
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4. A magician pretends to be able to allow an object, let's say a rabbit, float in the air with the wave of a wand. This is impossible because of Newton's laws. Use information from the text to explain what is really going on. Show your answer in words and an illustration.

2. Use Newton's Third Law of Motion and information from the text to draw a picture that explains why airplanes do not fall out of the sky.



3. If two students are running down the hall toward each other, trying to get to class, and they have the same mass and acceleration, what will happen when they collide? Will their forces cancel out or will each one experience a reaction? Use words and a diagram or picture to explain your reasoning.



Day 8

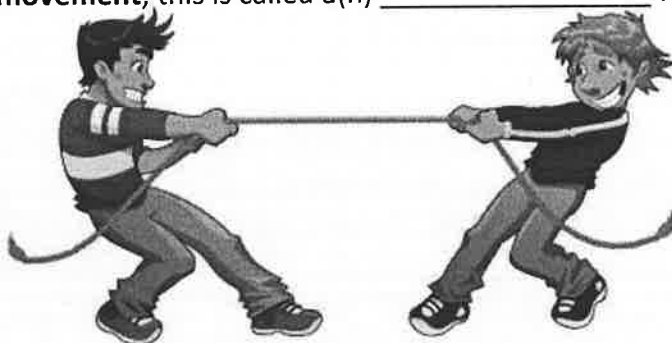
7th/8th Grades

Name: _____

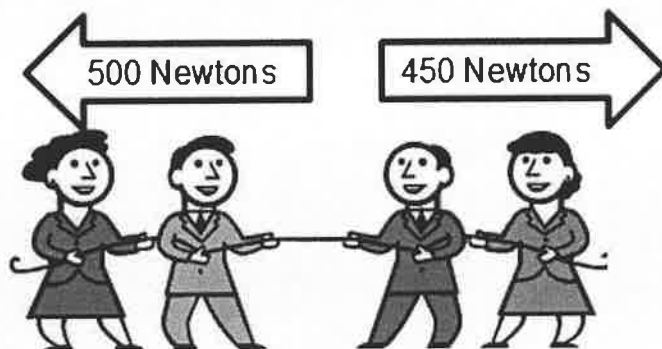
Balanced VS. Unbalanced Forces

A **force** can be a _____ or a _____. In the pictures below we see people exerting a force in one of these ways.

Sometimes there is **movement**, we called this a(n) _____ force.
Other times there is **no movement**, this is called a(n) _____ force.

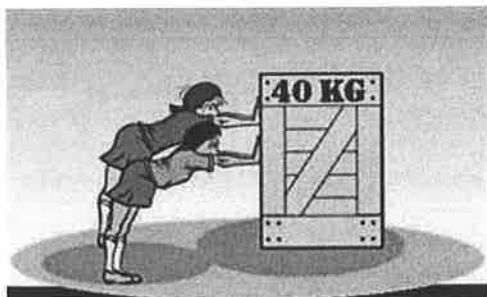


1. Is the picture above an example of pulling or pushing? _____
2. If Joey is exerting a force of 290N and Nick exerts 290N, who will win? _____
3. What is the total force applied to the rope? _____
4. Is this an example of a balanced or unbalanced force? _____



5. Above we have some teachers playing tug of war after school! Which side, left or right, will win? _____. By how much will they win? _____
6. Is this an example of a balanced or unbalanced force? _____

7. Below Sarah and Anita are pushing a crate. They both use 60N of force and the crate slowly slides. What is the total force on the crate? _____ It is *balanced/unbalanced*.

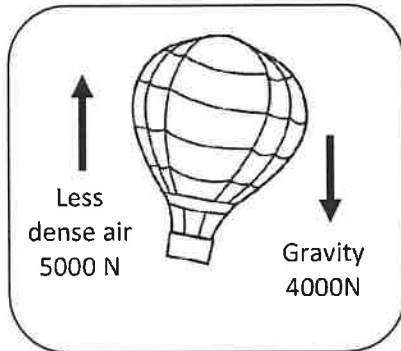


8. **WHAT IF** they both pushed on the crate and it didn't move?
What would the total force on the crate be then? _____
Now it would be considered *balanced/unbalanced*.

Day 9 7th/8th Grades

CALCULATING NET FORCE KEY

For each example, (1) identify the direction (same or opposite) that the main forces are acting on the object, (2) the method to calculate net force (add or subtract), (3) calculate the net force, and (4) identify if forces are balanced or unbalanced. Be sure to include your unit, Newtons (N).

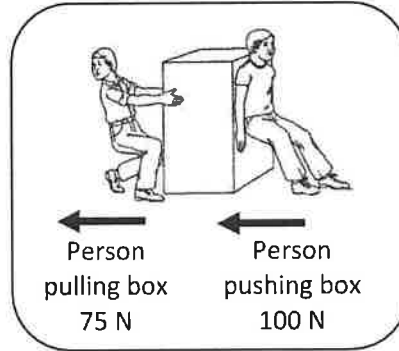


1.

2. subtract forces

3. $5000\text{N} - 4000\text{N} = 1000\text{N}$

4. Unbalanced

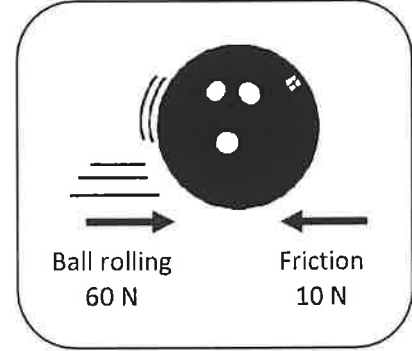


1. same directions

2. add forces

3. $75\text{N} + 100\text{N} = 175\text{N}$

4.

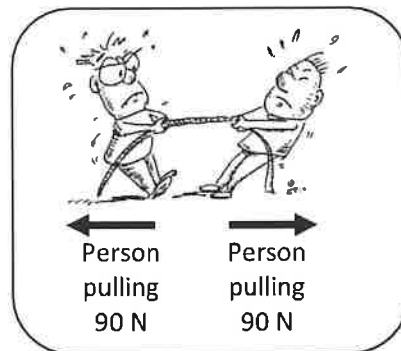


1. opposite directions

2.

3.

4. Unbalanced

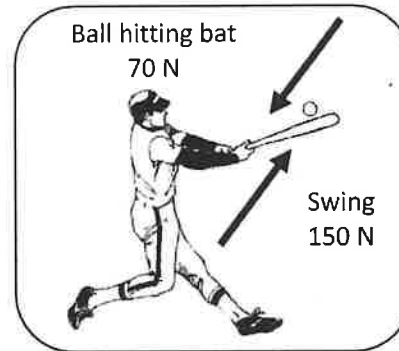


1.

2. subtract forces

3. $90\text{N} - 90\text{N} = 0\text{N}$

4. Balanced

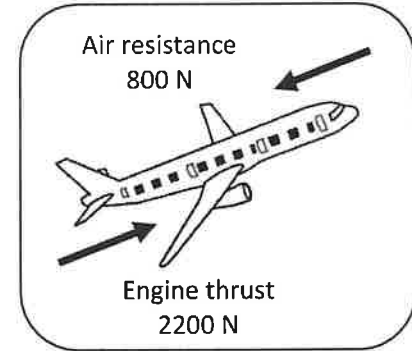


1. opposite directions

2. subtract forces

3.

4. Unbalanced



1. opposite directions

2. subtract forces

3. $2200\text{N} - 800\text{N} = 1400\text{N}$

4.

4th & 8th
Grade Day 10



Make a list of all the things shown in the above picture that are not safe lab practices.